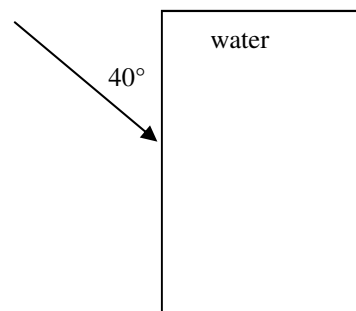


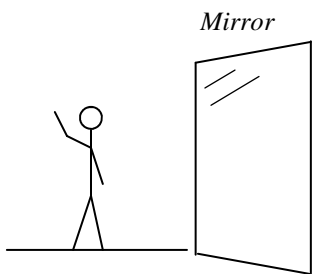
PreAP Light and Optics 5

- A substance has an index of refraction of 2.
 - * Calculate the speed of light in that substance.
 - How does the speed of light in the substance compare with that of the speed of light in a vacuum?
- So, (quickly, now), light travels 1×10^8 m/s in a substance. What is its index of refraction?
- 450 nm light traveling in air then passes into a tray of water, as shown. This time, I will walk you thru this.

- What part of the light wave is the same as it passes into water?
- * Calculate the frequency of the light in air.
- * Calculate the speed of light in the water (find n for water on the "Refraction" notes).
- * Calculate the frequency of the light in the water.
- Calculate the wavelength of the light in the water.
- Sketch the path of the light as it enters and exits the water tray.
- Calculate the angle that the light refracts in the water.



- At what angle will the light reflect off the surface?



- Slim Jim is waving hello to you. (He's a good guy!) Just so happens that he is standing next to a mirror. Draw the image of Jim you see in the mirror. (Think about what you see in your mirror at home.)

Let's start using a new equation...

- * A light wave has a frequency of 4×10^{15} Hz. How much energy does each photon have?
- * Photon I has a wavelength of 350 nm. Use $v = f\lambda$, solve for frequency, substitute into the formula and solve for energy of the photon.

Energy of one Photon		Planck's Constant ($6.63 \times 10^{-34} \text{ J}\cdot\text{s}$)
Energy per photon (in J/ photon)	$\rightarrow E = hf$	Photon's Frequency (in Hz)

- Photon II has a wavelength of 700 nm. How much energy is Photon II?

- Photon I or Photon II had more energy?
- Which of the following photons would have more energy?
 - Long wavelength or short wavelength?
 - High frequency or low frequency?
 - Blue or red light? (See "Light" notes)
 - Photon III or Photon IV at the right?

